STEREOSPECIFIC ACCESS TO BRANCHED-CHAIN CARBOHYDRATE SYNTHONS.

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Abstract: Appropriate precursors for the synthesis of D-vancosamine and L-evernitrose have been stereospecifically obtained from methyl 4,6-0-benzylidene-2-deoxy-X- and \$\beta\$-D-threo-hexopyranosid-3-ulose.

We have shown 1 that the addition of KCN to methyl 4,6-0-benzylidene-2-deoxy- α -D-erythro-hexopyranosid-3-ulose $\underline{1}$ under conditions of thermodynamic control in $\mathrm{CH}_2\mathrm{Cl}_2$ solution in the presence of NaHCO $_3$ and water, followed by mesylation, furnished a product $\underline{3}$ belonging to the arabino-series. In contrast, the addition of HCN to $\underline{1}$, under kinetic control in pyridine solution, followed by mesylation, has been shown by Yoshimura and al. 2 and by us 1 to afford a compound 2 with the ribo-configuration.

In an attempt to synthesize D-vancosamine $\frac{4}{4}$, the enantiomer of the branched-chain amino-sugar present in the glycopeptide antibiotic vancomycin 3 and L-evernitrose $\frac{5}{2}$, the naturally occurring nitro-sugar found in the oligosaccharide antibiotics evernino-micin-B, C and D 4 , methyl 4 ,6-0-benzylidene-2-deoxy- α -D-threo-hexopyranosid-3-ulose $\frac{5}{2}$ was submitted to the above mentioned reactions.Under conditions of thermodynamic control $\frac{5}{2}$ afforded a single cyano-mesylate $\frac{7}{2}$, yield $\frac{78}{2}$ [m.p. 97-99°C, α] $\frac{124}{2}$ (c 0.7) of unexpected configuration $\frac{1}{2}$.Under conditions of kinetic control, again a single cyano-mesylate $\frac{3}{2}$ was obtained, yield $\frac{73}{2}$ [m.p. $\frac{128-130}{2}$ C, α] $\frac{1}{2}$ + $\frac{150}{2}$ (c 1.3) whose configuration at C-3 was also considered as surprising in the light of previous results $\frac{1}{2}$.

These observations have prompted us to investigate the influence of the anomeric configuration on the steric course of the reactions. Thus, methyl 4,6-0-benzylidene-2-deoxy- β -D-three-hexopyranosid-3-ulose $\frac{9}{2}$ [m.p. 177-178°C [α] - 29° (c 1.0)] prepared in

80% yield by CrO_3 -pyridine oxidation of 10 was submitted to the cyanohydrin formation reactions. Each reaction afforded again a single cyano-mesylate. The technique which permitted to obtain 7 from 6 afforded 11 (yield 69%) [m.p. 140-141°C, $[\alpha]_D$ + 9°(c 1.3)] from 9 and the method which allowed the isolation of 8 from 6 gave 12 (yield 84%) [m.p. 127-128°C, $[\alpha]_D$ + 20°(c 1.3)] from 9. These results indicate that other factors than steric hindrance originating from the anomeric substituent are responsible for the stereochemistry at C-3 in the cyano-mesylates prepared.

Using the method of Bourgeois 7 , compounds 7 and 8 were transformed into the amines 7 [oil, [α]_D + 100°(c 1.0)] and 7 [α]_D + 76~79°C, [α]_D + 110°(c 0.7)] respectively with overall yields of 13% and 19%. In the same way, 11 and 12 were transformed into the amines 15 [oil, [α]_D - 21°(c 1.0)] and 16 [oil, [α]_D - 22°(c 1.0)] respectively with overall yields of 18% and 32%. Compounds 13 and 15 on the one hand and compound 14 on the other are appropriate precursors for the synthesis of the branched-chain sugars D-vancosamine 14 and L-evernitrose 5 respectively.

 13 C n.m.r. spectroscopy has been shown to be an excellent method to determine the configuration of substituents on quaternary centres 8 and this technique was used to establish the structure of all the new compounds. Carbon signal assignments for 7, 8, 11, 12, 13, 14, 15, 16 and for the models 17, 18, 19 and 20 are in the Table 9 . Model 17 was prepared from 21 by peracid treatment and AlLiH, reduction of the resulting two epoxides. The tertiary alcohols 17 [oil.[\mathbf{p}] $_0$ + 91° (c 1.1)] and 18 obtained in a ratio of 1: 4 were separated by preparative thin layer chromatography. Models 19 and 20 were obtained in a ratio of 6: 4 by a Grignard reaction from 9 with methyl magnesium iodide in ether. Attempts to separate them was unsuccessful and the spectral data shown in the Table result from an analysis of their mixture.

Table.	^{13}c	chemical	shifts	9	,a.

	7	8	11	12	<u>13</u>	14	<u>15</u>	<u>16</u>	<u>17</u>	18	<u>19</u>	20
C-1	95.9	97.3	97.2	99.0	99.2	99.7	100.9	99.9	99.0	99.7	100.9	99.9
C-2	32.6	34.1	36.3	36.0	39.1	36.9	41.2	39.6	39.1	35.8	41.3	39.4
C-3	74.2	74.6	77.6	76.4	48.8	49.2	54.2	52.0	70.0	68.9	71.4	69.7
C-4	72.3	73.4	71.9	72.8	80.0	80.5	79.1	78.8	78.9	78.4	78.2	77.8
C5	58.0	60.7	64.7	65.4	61.4	59.6	66.2	65.8	61.7	59.9	66.0	65.6
c-6	69.1	69.5	69.3	69.3	70.3	70.5	69.8	70.3	70.0	70.3	69.7	70.1
C≆N	115.0	116.1	114.6	115.6	-	-	-	-	_	_	_	-
OMe	55•4	55.5	56.7	57.0	55.0	55.3	56.4	56.3	55.0	55.4	56.5	56.5
SMe	40.1	40.6	40.4	40.3	-	-	-	-	_	_	_	-
C-3-Me	-	-	-	-	26.1	26.8	24.1	29.6	24.9	25.3	23.3	26.8

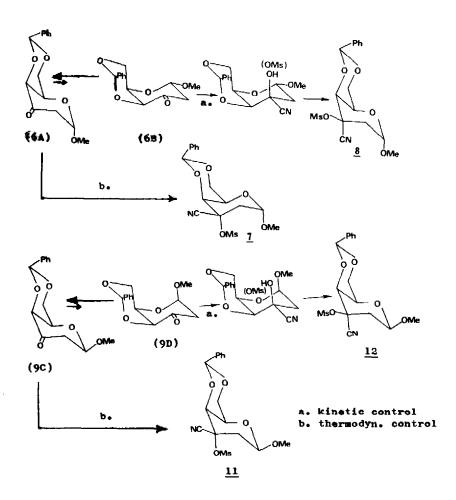
benzylidene carbon signals appear for all compounds at $101.5 \stackrel{+}{-} 0.5$, $2 \times 126.0 \stackrel{+}{-} 0.2$, $2 \times 128.3 \stackrel{+}{-} 0.2$, $129.1 \stackrel{+}{-} 0.2$ and $136.5 \stackrel{+}{-} 0.3$.

Structural differentiation for the epimeric compounds was based on the diagnostic C-5 chemical shift in the case of the \mathbf{K} -D-pyranosides and on the anomeric carbon shift for the $\boldsymbol{\beta}$ -D-pyranosides. It has been shown previously 1 that when C-5 bears an hydrogen atom situated 1,3-diaxially with respect to a C-3/O or C-3/N linkage, as in $\underline{7}$, $\underline{14}$ or $\underline{18}$, the chemical shift of C-5 is at higher field than in $\underline{8}$, $\underline{13}$ and $\underline{17}$, in which this hydrogen atom is 1,3-diaxially situated with respect to a C-3/CH $_3$ or C-3/CN bond. The same argument was excended to the anomeric carbon shifts in the case of the $\boldsymbol{\beta}$ -D-pyranosides $\underline{11}$, $\underline{12}$, $\underline{15}$, $\underline{16}$, $\underline{19}$ and $\underline{20}$.

In order to explain our apparently anomalous results under kinetic control, a conformational equilibrium between (6A) and (6B) was considered for the cis-decaline system. The equatorial approach of CN^- would be then governed by the axial orientation of the C-5/C-6 bond of conformer (6B) affording after mesylation the kinetic product G. In the case of G, where the conformational equilibrium between G0 and G1 should be strongly in favour of the former, the axial configuration of the G-G1 and G1 linkages on conformer G1 directs the equatorial approach of G1 thus furnishing G1.

Under thermodynamic control, reaction of CN on $\underline{6}$ and $\underline{9}$, followed by mesylation, permitted to isolate respectively $\underline{7}$ and $\underline{11}$ as the only stable compounds. Their stability may be related to the trans diaxially oriented two oxygen atoms attached to C-3 and C-4.

In the light of this interpretation, the structure of the products obtained in the cyanohydrin formation reactions from 1^{-1} , 6 and 9 is well understood.



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